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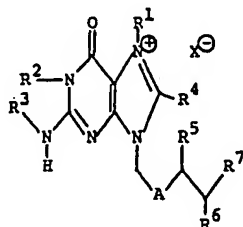
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⑤④ **N-alkylguanine acyclonucleosides as antiviral agents.**

⑤⑦ Disclosed are compounds of the formula:



(I)

—CH<sub>2</sub>OPO<sub>2</sub>OPO<sub>2</sub>O<sup>-</sup>—, or —OPO<sub>2</sub>OPO<sub>2</sub>O<sup>-</sup>—; A is O, S or CH<sub>2</sub> and X is a pharmaceutically acceptable anion. The compounds have antiviral activity, especially against viruses of the herpes class.

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and the pharmaceutically acceptable salts thereof wherein R<sup>1</sup> and R<sup>2</sup> are independently alkyl, haloalkyl, alkenyl, haloalkenyl, alkynyl or haloalkynyl, each having 1 to 19 carbon atoms, or R<sup>2</sup> is hydrogen; R<sup>3</sup> is hydrogen, alkyl having 1 to 6 carbon atoms or hydroxyalkyl having 1 to 6 carbon atoms; R<sup>4</sup> is hydrogen, halogen, amino or alkyl having 1 to 4 carbon atoms; R<sup>5</sup>, R<sup>6</sup> and R<sup>7</sup> are independently selected from hydrogen, hydroxy, alkyl having 1 to 6 carbon atoms, acyloxy having 1 to 8 carbon atoms, alkoxy having 1 to 6 carbon atoms, hydroxyalkyl having 1 to 6 carbon atoms, acyloxyalkyl having 1 to 12 carbon atoms, amino, alkylamino of 1 to 6 carbon atoms and —PO<sub>3</sub><sup>-</sup>, or two of R<sup>6</sup>, R<sup>6</sup> and R<sup>7</sup> taken together form a group —OPO<sub>2</sub>O<sup>-</sup>—, —CH<sub>2</sub>OPO<sub>2</sub>O<sup>-</sup>—,

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TITLE OF THE INVENTION:

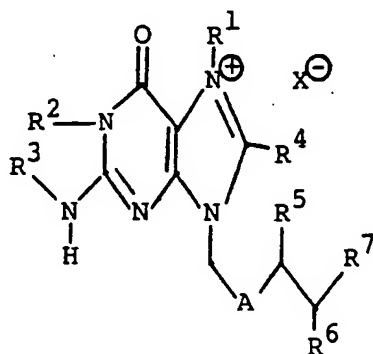
N-ALKYLGUANINE ACYCLONUCLEOSIDES AS  
ANTIVIRAL AGENTS

5           The present invention relates to  
N-alkylguanines. These compounds have antiviral  
activity. The compounds are particularly effective  
against herpes viruses, e.g. herpes simplex virus.  
The present invention also relates to processes for  
10       preparing said compounds, pharmaceutical compositions  
comprising said compounds and the treatment of viral  
infections in mammals with said compounds.

          The compounds of the present invention may  
be represented by the formula:

15

20



I

25

and the pharmaceutically acceptable salts thereof wherein  $R^1$  and  $R^2$  are independently alkyl, haloalkyl, alkenyl, haloalkenyl, alkynyl or haloalkynyl, each having 1 to 19 carbon atoms ( $R^1$  is preferably alkyl or alkenyl and more preferably methyl), or  $R^2$  is hydrogen;  $R^3$  is hydrogen, alkyl having 1 to 6 carbon atoms or hydroxyalkyl having 1 to 6 carbon atoms;  $R^4$  is hydrogen, halogen, amino or alkyl having 1 to 4 carbon atoms;  $R^5$ ,  $R^6$  and  $R^7$  are independently selected from hydrogen, hydroxy, alkyl having 1 to 6 carbon atoms, acyloxy having 1 to 8 carbon atoms, alkoxy having 1 to 6 carbon atoms, hydroxyalkyl having 1 to 6 carbon atoms, acyloxyalkyl having 1 to 12 carbon atoms, amino, alkylamino having 1 to 6 carbon atoms and  $-PO_3^-$  or two of  $R^5$ ,  $R^6$  and  $R^7$  taken together form a group  $-OPO_2O^-$ ,  $-CH_2OPO_2O^-$ ,  $-CH_2OPO_2OPO_2O^-$ , or  $-OPO_2OPO_2O^-$ ; A is O, S or  $CH_2$  and X is a pharmaceutically acceptable anion (preferably halide, alkanoate having 1 to 6 carbon atoms, alkylsulfonate having 1 to 6 carbon atoms, sulfate or phosphate). When the side chain at the 9-position on the guanine ring contains a strongly acidic monoanionic function (for example, a cyclic phosphate), that compound of the present invention will exist as a zwitterion, i.e., the compound will not require an accompanying anion. For example, the positive charge of the guaninium of 9-(2,2-dioxo-1,3,2-dioxaphosphorinan-5-yloxymethyl)-1,7-dimethylguanine is internally compensated for by the negative charge on the cyclic phosphate. The aforementioned alkyl groups, or the alkyl moieties of

other groups, may be linear, branched or cyclic or may contain both cyclic and linear or cyclic and branched moieties. Halogen includes fluorine, chlorine, bromine and iodine.

5 Preferred compounds of the present invention are compounds of the formula I wherein  $R^1$  and  $R^2$  are methyl,  $R^3$  and  $R^4$  are H,  $R^5$  is H or hydroxymethyl,  $R^6$  is H and  $R^7$  is hydroxyl or hydroxymethyl or, alternately,  $R^5$  and  $R^7$  taken  
10 together are  $-CH_2OPO_2O^-$ .

The following are representative compounds of the present invention:

- 9-(1,3-Dihydroxy-2-propoxymethyl)-1,7-dimethyl-guaninium iodide;
- 15 9-(1,3-Dihydroxy-2-propoxymethyl)-1-methyl-7-ethyl-guaninium iodide;
- 9-(1,3-Dihydroxy-2-propoxymethyl)-1-ethyl-7-methyl-guaninium iodide;
- 9-(1,3-Dihydroxy-2-propoxymethyl)-1-propyl-7-methyl-guaninium iodide;
- 20 9-(1,3-Dihydroxy-2-propoxymethyl)-1-(prop-2-ynyl)-7-methyl-guaninium iodide;
- 9-(1,3-Diacetoxy-2-propoxymethyl)-1,7-dimethyl-guaninium iodide;
- 25 9-(1,3-Di-n-octanoyloxy-2-propoxymethyl)-1,7-dimethyl-guaninium iodide;
- 9-(1,3-Dihydroxy-2-propoxymethyl)-1-(prop-2-enyl)-7-methyl-guaninium iodide;
- 9-(1,3-Dihydroxy-2-propoxymethyl)-1,7-dimethyl-guaninium acetate;
- 30 9-(2,3-Dihydroxy-1-propoxymethyl)-1,7-dimethyl-guaninium iodide;

- 9-(2-Hydroxyethoxymethyl)-1,7-dimethylguaninium  
iodide;
- 9-(4-Hydroxybutyl)-1,7-dimethylguaninium iodide;
- 5 9-(4-Hydroxy-3-hydroxymethylbutyl)-1,7-dimethyl-  
guaninium iodide;
- 9-(2-hydroxy-1,3,2-dioxaphosphorinan-5-yloxymethyl)-  
1,7-dimethylguanine P-oxide;
- 9-(1,3-Dihydroxy-2-propoxymethyl)-1-methyl-7-(prop-2-  
enyl)guaninium iodide;
- 10 9-(1,3-Dihydroxy-2-propoxymethyl)-1-methyl-7-(prop-2-  
ynyl)guaninium iodide;
- 9-(1,3-Dihydroxy-2-propoxymethyl)-1-methyl-7-(3-methyl-  
but-2-enyl)guaninium iodide;
- 9-(1,3-Dihydroxy-2-propoxymethyl)-1-methyl-7-(hex-2-  
enyl)guaninium iodide;
- 15 9-(1,3-Dihydroxy-2-propoxymethyl)-1-methyl-7-(but-3-  
ynyl)guaninium iodide;
- 9-(1,3-Dihydroxy-2-propoxymethyl)-1-methyl-7-ethynyl-  
guaninium iodide;
- 20 9-(1,3-Dihydroxy-2-propoxymethyl)-1-methyl-7-hexadecyl-  
guaninium iodide;
- 9-(1,3-Dihydroxy-2-propoxymethyl)-1-methyl-7-(oct-7-  
ynyl)guaninium iodide;
- 9-(2-Hydroxyethoxymethyl)-1-ethyl-7-methylguaninium  
chloride;
- 25 9-(2-Hydroxyethoxymethyl)-1-propyl-7-methylguaninium  
chloride;
- 9-(2-Hydroxyethoxymethyl)-1-ethenyl-7-methylguaninium  
chloride;
- 30 9-(2-Hydroxyethoxymethyl)-1-(prop-2-ynyl)-7-methyl-  
guaninium chloride;
- 9-(4-Hydroxybutyl)-1,7-dimethyl-8-aminoguaninium  
propanoate;

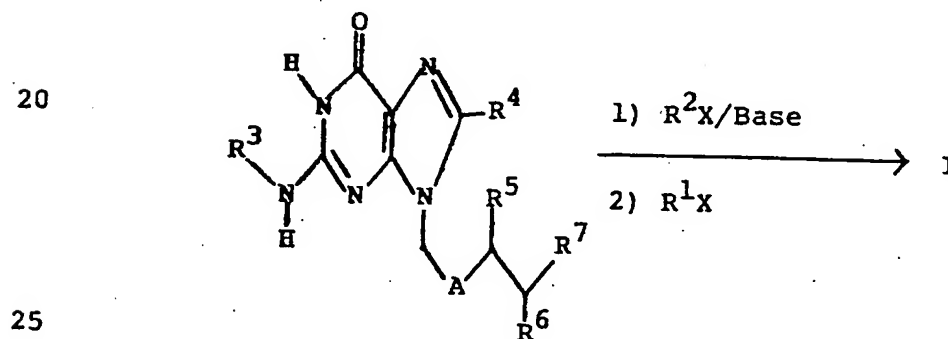
- 9-(4-Hydroxybutyl)-1,7-dimethyl-8-bromoguaninium  
propanoate;  
9-(4-Hydroxybutyl)-1,7-dimethyl-8-chloroguaninium  
propanoate;  
5 9-(4-Hydroxybutyl)-1,7,8-trimethyl-guaninium  
propanoate;  
9-(4-Hydroxybutyl)-1,7-dimethyl-N<sup>2</sup>-(2-hydroxyethyl)-  
guaninium propanoate;  
9-(4-Hydroxybutyl)-1,7-dimethyl-N<sup>2</sup>-(2,3-dihydroxy-  
10 propyl)guaninium propanoate;  
9-(3,4-Dihydroxybutyl)-1,7-dimethylguaninium ethyl-  
sulfonate;  
9-(3-Hydroxypropyloxymethyl)-1,7-dimethylguaninium  
ethylsulfonate;  
15 9-(2-Hydroxyethylthiomethyl)-1,7-dimethylguaninium  
ethylsulfonate;  
9-(2,4-Dihydroxy-1,3,5,2,4-trioxadiphosphhepan-6-  
yloxymethyl)-1,7-dimethylguanine ethylsulfonate  
P,P'-dioxide;  
20 9-(2,4-Dihydroxy-1,3,5,2,4-trioxadiphosphacan-7-  
yloxymethyl)-1,7-dimethylguanine ethylsulfonate  
P,P'-dioxide;  
9-(1-Hydroxy-3-methoxy-2-propoxymethyl)-1,7-dimethyl-  
guaninium phosphate;  
25 9-(1-Hydroxy-3-methylamino-2-propoxymethyl)-1,7-  
dimethylguaninium phosphate; and  
9-(1-Hydroxy-3-phosphoryloxy-2-propoxymethyl)-1,7-  
dimethylguanine.

30 The following compounds are preferred:

- 9-(1,3-Dihydroxy-2-propoxymethyl)-1,7-dimethyl-  
guaninium iodide;

- 9-(1,3-Dihydroxy-2-propoxymethyl)-1-methyl-7-ethyl-guaninium iodide;  
 9-(1,3-Dihydroxy-2-propoxymethyl)-1-ethyl-7-methyl-guaninium iodide;  
 5 9-(1,3-Dihydroxy-2-propoxymethyl)-1,7-dimethyl-guaninium acetate;  
 9-(4-Hydroxybutyl)-1,7-dimethylguaninium iodide;  
 9-(4-Hydroxy-3-hydroxymethylbutyl)-1,7-dimethyl-guaninium iodide;  
 10 9-(2-Hydroxy-1,3,2-dioxaphosphorinan-5-yloxymethyl)-1,7-dimethylguanine P-oxide; and  
 9-(1,3-Di-n-octanoyloxy-2-propoxymethyl)-1,7-dimethyl-guaninium iodide.

15 The compounds of the present invention may be prepared as shown in the following scheme:



II

30 As shown above, Compound II is alkylated at N<sup>1</sup> with a suitable alkylating agent (e.g. an alkyl halide) in the presence of one equivalent of base (e.g. NaH or K<sub>2</sub>CO<sub>3</sub>). This is followed by alkylation at N<sup>7</sup> at or near neutral pH with a

suitable alkylating agent such as an alkyl halide. Also, dialkylation can be achieved by alkylation at N<sup>7</sup>, first under neutral conditions, followed by alkylation at N<sup>1</sup> after the addition of 2  
5 equivalents of base. If R<sup>1</sup> and R<sup>2</sup> are identical, dialkylation may be carried out in a single step by reacting with two equivalents of a suitable alkylating agent (such as an alkyl halide) in the presence of base.

10 The above procedure is applicable to a wide range of substituted acyclonucleosides. For example, 2- and 8-substituted guanines are readily available by procedures known to those skilled in the art. Similarly, N-substituted guanines are readily  
15 available from protected guanines by general procedures employing various types of acyclonucleoside side chains.

For example, U.S. Serial No. 574,113, filed January 26, 1984, discloses an acyclonucleoside with  
20 a 4-hydroxy-3-hydroxymethylbutyl side chain. Also, using a preformed, protected, guanine acyclonucleoside, selective tosylation of hydroxyl groups on the side chain may be effected and nucleophilic displacement with substituted amines or alkoxides  
25 furnishes alkylamino or alkoxy substituted guanine acyclonucleosides. In addition, U.S. Serial No. 533,676, filed September 19, 1983, discloses cyclic pyrophosphates of purine acyclonucleosides. 2- and 8-haloguanine acyclonucleosides are readily available  
30 by acyclonucleoside synthesis using preformed halopurines or, in the case of 8-substitution, the halogen can also be introduced directly by electrophilic substitution. Other 8-substituted



guanine acyclonucleosides are prepared by nucleophilic substitution of 8-halo guanine derivatives, for example 8-amino, or by introduction of the 8-substituent into the purine moiety before  
5 alkylation by the side chain intermediate.

Pharmaceutically acceptable salts of the compound of the present invention may be prepared by ion-exchange chromatography from an appropriate salt (for example, the iodide, chloride or acetate salt)  
10 and the appropriate anion-exchange resin.

In another aspect of the invention there is provided a pharmaceutical composition or preparation comprising a compound of the formula I, or a pharmaceutically acceptable salt thereof, together  
15 with a pharmaceutically acceptable carrier therefor. In a particular aspect the pharmaceutical composition comprises a compound of the present invention in effective unit dosage form.

As used herein the term "effective unit  
20 dosage" or "effective unit dose" is denoted to mean a predetermined antiviral amount sufficient to be effective against the virus in vivo. Pharmaceutically acceptable carriers are materials useful for the purpose of administering the medicament, and  
25 may be solid, liquid or gaseous materials, which are otherwise inert and medically acceptable and are compatible with the active ingredients.

These pharmaceutical compositions may be given parenterally, orally, used as a suppository or  
30 pessary, applied topically as an ointment, cream, aerosol, powder, or given as eye or nose drops, etc., depending on whether the preparation is used to treat internal or external viral infections.

For internal infections the compositions are administered orally or parenterally at dose levels of about 0.1 to 250 mg per kg, preferably 1.0 to 50 mg per kg of mammal body weight, and are used in man in a unit dosage form, administered, e.g. a few times daily, in the amount of 1 to 250 mg per unit dose.

For oral administration, fine powders or granules may contain diluting, dispersing and/or surface active agents, and may be presented in a draught, in water or in a syrup; in capsules or sachets in the dry state or in a non-aqueous solution or suspension, wherein suspending agents may be included; in tablets, wherein binders and lubricants may be included; or in a suspension in water or a syrup. Where desirable or necessary, flavoring, preserving, suspending, thickening or emulsifying agents may be included. Tablets and granules are preferred, and these may be coated.

For parenteral administration or for administration as drops, as for eye infections, the compounds may be presented in aqueous solution in a concentration of from about 0.1 to 10%, more preferably 0.1 to 7%, most preferably 0.2% w/v. The solution may contain antioxidants, buffers, etc.

Alternatively, for infections of the eye, or other external tissues, e.g. mouth and skin, the compositions are preferably applied to the infected part of the body of the patient as a topical ointment or cream. The compounds may be presented in an ointment, for instance, with a water soluble ointment base, or in a cream, for instance with an oil in water cream base, in a concentration of from about 0.1 to 10%, preferably 0.1 to 7%, most preferably 1% w/v.

The compounds of the present invention may also be administered in combination with other antiviral drugs such as acyclovir. Because the compounds of the present invention are not converted to the corresponding triphosphate in virus-infected cells and conversion to the triphosphate is not important for expression of antiviral activity as are other nucleoside antiviral agents, the compounds of the present invention will form synergistic combinations with other antiviral agents.

The following examples illustrate the present invention without, however, limiting the same thereto. All temperatures are expressed in degrees Celsius.

#### EXAMPLE 1

##### 1-Methyl-9-(1,3-dihydroxy-2-propoxymethyl)guanine

To a stirred solution of 9-(1,3-dihydroxy-2-propoxymethyl)guanine (510.4 mg, 2.0 mmol) in sieve-dried DMSO (dimethylsulfoxide) (4 ml), under  $N_2$ , was added 80 mg of 60% NaH in oil (i.e. 48 mg of NaH, 2.0 mmol). Effervescence was observed and after 10 minutes a clear solution was obtained. Methyl iodide (312 mg, 2.20 mmol) in dry DMSO (dimethylsulfoxide) (1 ml) was added in 3 portions over a period of 5 minutes. After stirring overnight at room temperature the reaction mixture was poured into  $CH_2Cl_2$  (200 ml) and the precipitate so formed was filtered off. This was dissolved in 15 ml of MeOH- $H_2O$  (1:4) and applied to an ion-exchange column of Dowex 1 X 2 ( $OH^-$  form, 3.5 X 18.5 cm) packed in the same solvent. The column was developed with MeOH- $H_2O$  (1:4) and fractions containing the

required product were pooled and evaporated to dryness. The white powder so obtained (350 mg, 1.30 mmol; 65%) had a melting point of 222-222.5°C and was analytically pure.

5 Anal.: Calcd. for  $C_{10}H_{15}N_5O_4$ :

C, 44.61; H, 5.62; N, 26.01.

Found: C, 44.23; H, 5.64; N, 25.69.

UV (MeOH):  $\lambda_{\max}$  255 nm ( $\epsilon=10,320$ ), shoulder 270 nm;

(0.01M HCl):  $\lambda_{\max}$  255 nm ( $\epsilon=9,200$ ), shoulder 270

10 nm; (0.01M NaOH):  $\lambda_{\max}$  252 nm ( $\epsilon=10,000$ ), shoulder 265 nm.

$^{13}\text{C}$ MR and PMR were in agreement with the structure.

#### EXAMPLE 2

##### 15 1-Ethyl-9-(1,3-dihydroxy-2-propoxymethyl)guanine

To a stirred solution of 9-(1,3-dihydroxy-2-propoxymethyl)guanine (766 mg, 3.0 mmol) in sieve-dried DMSO (4 ml), under  $N_2$ , was added 120 mg of 60% NaH in oil (i.e. 72 mg NaH, 3.0 mmol).

20 Hydrogen evolution ceased and a clear solution was obtained after 10 minutes. Ethyl iodide (491 mg, 3.15 mmol) in DMSO (1 ml) was added over

approximately 1 minute. The reaction was stirred overnight and then poured into  $CH_2Cl_2$ . The gummy

25 precipitate was filtered off and triturated under methanol to give crystalline material. This was dissolved in MeOH- $H_2O$  (2:3) and applied to a Dowex 1 x 2 column (OH<sup>-</sup> form, 100 ml) packed in the same solvent. The column was developed in MeOH- $H_2O$

30 (2:3) and fractions containing the required product were pooled and evaporated to dryness. This residue was crystallized from methanol to give 230 mg (27% yield) of product.

Anal.: Calculated for  $C_{11}H_{17}N_5O_4$ :

C, 46.64; H, 6.05; N, 24.72

Found: C, 46.82; H, 6.07; N, 24.84

UV (MeOH):  $\lambda_{\max}$  257 nm ( $\epsilon=13,000$ ), shoulder 270 nm;

5 (0.01M HCl):  $\lambda_{\max}$  257 nm ( $\epsilon=11,074$ ), shoulder 275 nm;

(0.01M NaOH):  $\lambda_{\max}$  255 nm ( $\epsilon=12,230$ ), shoulder 270 nm;

<sup>13</sup>CMR and PMR were in agreement with the structure.

### EXAMPLE 3

10 1-n-Propyl-9-(1,3-dihydroxy-2-propoxymethyl)guanine

9-(1,3-Dihydroxy-2-propoxymethyl)guanine

(766 mg, 3.0 mmol) and 120 mg of 60% NaH in oil (i.e.

72 mg of NaH, 3.0 mmol) were stirred vigorously under

$N_2$  with dry DMSO (4 ml). After the evolution of

15  $H_2$  had ceased and a clear solution was obtained,

n-propyl iodide (535 mg, 3.15 mmol) was added and the reaction was stirred overnight at room temperature.

The mixture was then poured into  $CH_2Cl_2$  (250 ml)

and a gummy precipitate was formed which was filtered

20 off after standing for 1 hour. This was taken up in

aqueous MeOH and the precipitate so formed (unreacted

9-(1,3-dihydroxy-2-propoxymethyl)guanine, 115 mg) was

filtered off. The filtrate was concentrated to an

oil and applied to a Dowex 1x2 column ( $OH^-$  form)

25 packed in MeOH- $H_2O$  (15:85). The column was

developed first in MeOH- $H_2O$  (15:85) and then with

MeOH- $H_2O$  (3:7) and fractions containing the

required product were pooled and evaporated to

dryness to give 31% overall yield of product.

30 Analytically pure material was obtained by

crystallization from 2-propanol-MeOH.

Anal.: Calcd for  $C_{12}H_{19}N_5O_4 \cdot 0.8 H_2O$ :

C, 46.23; H, 6.66; N, 22.47

Found: C, 46.55; H, 6.53; N, 23.34

UV (MeOH):  $\lambda_{max}$  257 nm ( $\epsilon=14,280$ ), shoulder 270 nm;

5 (0.01M HCl):  $\lambda_{max}$  257 nm ( $\epsilon=12,000$ ), shoulder 275 nm;

(0.01M NaOH):  $\lambda_{max}$  255 nm ( $\epsilon=13,270$ ), shoulder 270 nm;

<sup>13</sup>CMR and PMR were in agreement with the structure.

#### EXAMPLE 4

10 7-Methyl-9-(1,3-dihydroxy-2-propoxymethyl)guanine  
iodide

To a stirred solution of 9-(1,3-dihydroxy-2-propoxymethyl)guanine (510 mg, 2.0 mmol) in sieve-dried DMF (dimethylformamide) (50 ml) was added  
15 a solution of methyl iodide (305 mg; 2.15 mmol) in dry DMF (2 ml). After stirring at room temperature for 5 hours, little reaction was apparent by TLC (thin layer chromatography) evaluation and the reaction was heated at 60° under a reflux condenser  
20 overnight. TLC then indicated complete reaction and the mixture was cooled and evaporated to dryness, giving an oil. This was evaporated twice to dryness from MeOH and a crystalline product was obtained. This material was recrystallized from MeOH (25 ml)  
25 and the product was filtered after standing 3 days at ambient temperature. The yield was 260 mg (0.65 mmol, 33%). An analytical sample was obtained by recrystallization from absolute EtOH.

Anal.: Calcd for  $C_{10}H_{16}N_5O_4I$ :

30 C, 30.24; H, 4.06; N, 17.63.

Found: C, 30.65; H, 4.13; N, 17.53.

UV (MeOH):  $\lambda_{max}$  222 nm ( $\epsilon=22,880$ ), 255 nm

( $\epsilon$ =6,100), 283 nm ( $\epsilon$ =6,390); (0.01M HCl):  $\lambda_{\max}$  256 nm ( $\epsilon$ =10,490), shoulder 275 nm.

<sup>13</sup>CMR and PMR were in agreement with the structure.

5

EXAMPLE 5

1-Ethyl-7-methyl-9-(1,3-dihydroxy-2-propoxymethyl)  
guanine iodide

1-Ethyl-9-(1,3-dihydroxy-2-propoxymethyl)  
guanine (259 mg, 0.91 mmol) and methyl iodide (142 mg,  
10 1.0 mmole) were dissolved in dry DMF (3 ml) and heated  
at 50° overnight. The reaction mixture was poured  
into CH<sub>2</sub>Cl<sub>2</sub> (230 ml) to give a cloudy solution  
which deposited solid on the walls of the flask after  
standing for 5 hours at 4°. The liquid was decanted  
15 off and the solid was triturated under CH<sub>2</sub>Cl<sub>2</sub> and  
then removed by centrifugation to give 261 mg of crude  
product. This was recrystallized from MeOH to give  
156 mg (54% yield) of analytically pure material  
20 having a melting point of 148-150°.

Anal.: Calcd for C<sub>12</sub>H<sub>20</sub>N<sub>5</sub>O<sub>4</sub>I<sub>1</sub>:

C, 33.89; H, 4.74; N, 16.47.

Found: C, 33.99; H, 4.75; N, 16.37.

UV (MeOH):  $\lambda_{\max}$  262 nm ( $\epsilon$ =10,880), shoulder 280 nm

25 (0.01 M HCl):  $\lambda_{\max}$  259 nm ( $\epsilon$ =10,000), shoulder 277 nm;

<sup>13</sup>CMR and PMR were in agreement with the structure.

EXAMPLE 6

1-Propyl-7-methyl-9-(1,3-dihydroxy-2-propoxy-  
30 methyl)guanine iodide

Following the method of Example 5, using  
1-propyl-9-(1,3-dihydroxy-2-propoxymethyl)guanine and  
methyl iodide in DMF at 60°C overnight, prepare  
1-propyl-7-methyl-9-(1,3-dihydroxypropoxymethyl)-  
guanine iodide.

EXAMPLE 7

1,7-Dimethyl-9-(1,3-dihydroxy-2-propoxymethyl)guanine  
iodide

Method A: To a stirred mixture of 9-(1,3-dihydroxy-2-  
5 propoxymethyl)guanine (1.0 g, 3.92 mmol) and dried  
 $K_2CO_3$  (1.0 g) in dry DMSO (4 ml) was added a  
solution of methyl iodide (1.0 g, 7.05 mmol) in dry  
DMSO (2 ml). The dropwise addition took 5 minutes.  
The reaction mixture was stirred at room temperature  
10 for 5 hours, filtered through Celite (diatomaceous  
earth) and was then poured into  $CH_2Cl_2$  (200 ml).  
The white solid so obtained (1.6 g) was  
recrystallized from MeOH (50 ml) and the product was  
filtered after standing overnight in the refrigerator  
15 (0.8 g, 1.95 mmol, 50%). A second recrystallization  
from MeOH was necessary to remove minute traces of  
starting material.

Melting point: sample softens at 165-170°, turns  
brown at 220-225° and finally melts with  
20 decomposition at 260-262°.

Anal.: Calcd. for  $C_{11}H_{18}N_5O_4I$ :

C, 32.13; H, 4.41; N, 17.03.

Found: C, 31.99; H, 4.36; N, 16.98.

UV (MeOH):  $\lambda$  max 261 nm ( $\epsilon$ =10,690), shoulder 275  
25 nm; (0.01M HCl):  $\lambda$  max 258 nm ( $\epsilon$ =12,130).

$^{13}C$ MR and PMR were in agreement with the structure.

Method B:

1-Methyl-9-(1,3-dihydroxy-2-propoxymethyl)  
30 guanine (164 mg; 0.61 mmol) and methyl iodide  
(100 mg, 0.7 mmol) were mixed with dry DMF (5 ml) and  
heated to 70° in a pressure bottle for 8 hours. The  
mixture was concentrated to an oil and  $CH_2Cl_2$



was added. A precipitate formed after trituration which was removed by centrifugation. This solid was crystallized from MeOH to give material identical to that prepared from Methods A and C.

5

Method C:

7-Methyl-9-(1,3-dihydroxy-2-propoxymethyl) guanine iodide (300 mg, 0.76 mmol), methyl iodide (216 mg, 1.52 mmol) and dry  $K_2CO_3$  (126 mg, 0.91 mmol) were stirred in dry DMSO (5 ml) at room temperature for 4 hours. The reaction was filtered and concentrated to an oil which was triturated under  $CH_2Cl_2$  (40 ml) to give a white precipitate. This crude product was crystallized from MeOH to give 160 mg of product identical to material prepared by Methods A and B.

EXAMPLE 8

1-Methyl-9-(2-hydroxyethoxymethyl)guanine

To a stirred solution of 9-(2-hydroxyethoxymethyl)guanine (500 mg; 2.22 mmol) in sieve-dried DMSO (4 ml), under  $N_2$ , was added 98 mg of 60% NaH in oil (i.e. 58.8 mg of NaH, 2.45 mmol). After the evolution of  $H_2$  had ceased, a clear solution was obtained after 15 minutes. Methyl iodide (315 mg, 2.22 mmol) in dry DMSO (1.5 ml) was added over a period of about 1 minute and the reaction mixture was stirred under  $N_2$  at room temperature overnight. The mixture was added to  $CH_2Cl_2$  (200 ml) and the crude product formed a gum. The supernatant was decanted (some solid material was filtered and then mixed back with the gum) and the gum was dissolved in 20 ml of MeOH- $H_2O$

(1:4) and applied to an ion-exchange column of Dowex 1 X 2 (OH<sup>-</sup> form, 3.5 x 19 cm) packed in the same solvent. The column was developed with MeOH-H<sub>2</sub>O (1:4) and fractions containing the required product were pooled and evaporated to dryness (yield, 250 mg, 1.05 mmol, 47%). This material was crystallized from MeOH (about 150 ml) to give 201 mg of analytically pure material having a melting point of 235-236°.

Anal.: Calcd. for C<sub>9</sub>H<sub>13</sub>N<sub>5</sub>O<sub>3</sub>:

10 C, 45.18; H, 5.48; N, 29.28.

Found: C, 45.10; H, 5.48; N, 29.04.

UV (MeOH):  $\lambda_{\text{max}}$  256.5 nm ( $\epsilon$ =11,310); (0.01 M HCl):  $\lambda_{\text{max}}$  256.5 nm ( $\epsilon$ =10,660); (0.01M NaOH):  $\lambda_{\text{max}}$  254.5 nm ( $\epsilon$ =11,200).

15 <sup>13</sup>CMR and PMR were in agreement with the structure.

#### EXAMPLE 9

##### 7-Methyl-9-(2-hydroxyethoxymethyl)guanine iodide

To a stirred solution of 9-(2-hydroxyethoxy-methyl)guanine (1.0 g; 4.44 mmol) in dry DMF (50 ml) was added a solution of methyl iodide (680 mg, 4.77 mmol) in dry DMF (2 ml). This mixture was heated under a reflux condenser under N<sub>2</sub> at 57°C overnight. The mixture was concentrated in vacuo to an oil and the evaporation was repeated several times from MeOH. The residue was dissolved in MeOH (20 ml) and 2-propanol (150 ml) was added and the mixture was stirred overnight. A yellow solid was obtained which was filtered off (200 mg). This was recrystallized from MeOH (25 ml) (solution filtered through a little charcoal). Crystallization was induced by concentration of the solution, cooling and by the addition of a little 2-propanol.

EXAMPLE 101,7-Dimethyl-9-(2-hydroxyethoxymethyl)guanine iodide

1.0 g (4.44 mmole) of 9-(2-hydroxyethoxy-  
methyl)guanine was dissolved in sieve-dried DMSO (4  
5 ml) and anhydrous  $K_2CO_3$  (1.35 g; 9.77 mmol) was  
added. To this stirred mixture was added methyl  
iodide (1.40 g; 9.86 mmol) in dry DMSO (2 ml) over a  
15 minute period. After stirring overnight at room  
temperature, the mixture was filtered through a  
10 Celite pad. The filtrate was diluted to 400 ml with  
 $CH_2Cl_2$  and the white precipitate so formed was  
filtered off to give the crude product. This was  
recrystallized twice from MeOH to give 711 mg of pure  
product (42%) with a melting point of 255-256°  
15 (decomp.; softens at 240-250°).

Anal.: Calculated for  $C_{10}H_{16}N_5O_5I$ :

C, 31.51; H, 4.23; N, 18.37.

Found: C, 31.49; H, 4.21; N, 18.17.

UV (MeOH):  $\lambda_{max}$  262 nm ( $\epsilon=12,310$ ), shoulder 280  
20 nm; (0.01M HCl):  $\lambda_{max}$  258 nm ( $\epsilon=11,370$ ), shoulder  
275 nm.

13CMR and PMR were in agreement with the structure.

EXAMPLE 11(S)-1,7-Dimethyl-9-(2,3-dihydroxy-1-propoxymethyl)-  
guanine iodide

0.500 g (1.96 mmol) of (S)-9-(2,3-dihydroxy-  
1-propoxymethyl)guanine was dissolved in sieve-dried  
DMSO (4 ml) and powdered anhydrous  $K_2CO_3$  (0.677 g;  
30 4.9 mmol) was added. To this stirred mixture was  
added methyl iodide (0.700 g; 4.9 mmol) in dry DMSO  
(2 ml) in one portion. After stirring overnight at  
room temperature, the reaction mixture was filtered

through a Celite pad, washing with 2 ml of DMSO. The filtrate was diluted with  $\text{CH}_2\text{Cl}_2$  (400 ml) and the white precipitate so formed was filtered off after standing at room temperature. The product was  
5 recrystallized from 10 ml MeOH (filtered after chilling to  $4^\circ$ ) to give 0.42 g of product having a melting point of  $143-145^\circ$  (decomp.).

UV (MeOH):  $\lambda_{\text{max}}$  261 nm ( $\epsilon=11,990$ ), shoulder 280 nm; (0.01 M HCl):  $\lambda_{\text{max}}$  258 nm ( $\epsilon=11,140$ ), shoulder  
10 275 nm.

$^{13}\text{C}$ MR and PMR were in agreement with the structure.

Anal.: Calculated for  $\text{C}_{11}\text{H}_{18}\text{N}_5\text{O}_4 \cdot 0.6\text{H}_2\text{O}$ :

C, 31.31; H, 4.56; N, 16.60.

Found: C, 31.62; H, 4.48; N, 16.17.

15

#### EXAMPLE 12

##### 1-Methyl-9-(1,3-dioctanoyloxy-2-propoxymethyl)guanine

1-Methyl-9-(1,3-dihydroxy-2-propoxymethyl)guanine (340 mg, 1.26 mmol) was suspended in dry DMF  
20 and dry pyridine (approximately 20 ml total) and evaporated to dryness. This process was repeated twice, the final time concentrating the suspension down to 10 ml. This suspension was cooled to  $0^\circ$ , under  $\text{N}_2$ , and a solution of octanoyl chloride (822  
25 mg, 5.05 mmol) in dry DMF (1 ml) was added. This reaction was stirred overnight at room temperature. Methylene chloride was then added and the mixture was extracted with saturated aqueous  $\text{NaHCO}_3$  solution. The organic phase was then washed three times with  
30  $\text{H}_2\text{O}$ , dried over  $\text{MgSO}_4$ , filtered and evaporated to dryness. The residual oil was dissolved in  $\text{CH}_2\text{Cl}_2$  and applied to a column of silica gel, packed in  $\text{CH}_2\text{Cl}_2$ . Elution was first performed

with  $\text{CH}_2\text{Cl}_2$  followed by 1% MeOH in  $\text{CH}_2\text{Cl}_2$  (200 ml), 2% MeOH in  $\text{CH}_2\text{Cl}_2$  (200 ml), 3% MeOH in  $\text{CH}_2\text{Cl}_2$  (100 ml) and finally 5% MeOH in  $\text{CH}_2\text{Cl}_2$  (100 ml). Fractions containing the required product were pooled and evaporated to dryness to give 529 mg of product. It was recrystallized from ether/petroleum ether. The PMR spectrum was in accord with the structure.

Anal.: Calculated for  $\text{C}_{26}\text{H}_{43}\text{N}_5\text{O}_6$ :

10 C, 59.86; H, 8.31; N, 13.43.

Found: C, 59.86; H, 8.27; N, 13.51.

UV(MeOH):  $\lambda_{\text{max}}$  257 nm ( $\epsilon=12,860$ ), shoulder 269 nm

#### EXAMPLE 13

15 1,7-Dimethyl-9-(1,3-diocanoyloxy-2-propoxymethyl)  
guanine iodide

##### Method A:

9-(1,3-diocanoyloxy-2-propoxymethyl)guanine (200 mg, 0.394 mmol) and anhydrous  $\text{K}_2\text{CO}_3$  (114 mg, 0.827 mmol) were mixed in dry DMSO (2 ml) and stirred at room temperature. To this mixture was added methyl iodide (117 mg, 0.827 mmol) and the reaction was heated at 50° overnight. Additional methyl iodide (excess) was then added and the mixture was heated at 70° in a pressure tube overnight. The reaction mixture was filtered, evaporated to dryness and the residue was dissolved in  $\text{CHCl}_3$  and applied to a silica gel column. The column was developed first with  $\text{CHCl}_3$ -MeOH- $\text{H}_2\text{O}$  (95:5:0.5) and then with  $\text{CHCl}_3$ -MeOH- $\text{H}_2\text{O}$  (90:10:1). Fractions containing the required product were pooled and evaporated to dryness to give 50 mg of chromatographically pure product. This residue was

partitioned between  $\text{CHCl}_3$  and  $\text{H}_2\text{O}$  and the organic phase was dried over  $\text{MgSO}_4$ , filtered and evaporated to dryness. The residue was crystallized from  $\text{CHCl}_3$ -ethyl ether to give 26 mg of analytically pure product.

Anal.: Cal'd for  $\text{C}_{27}\text{H}_{46}\text{N}_5\text{O}_6\text{I}$ :

C, 48.86; H, 6.98; N, 10.55.

Found: C, 48.91; H, 7.03; N, 10.51.

UV(MeOH):  $\lambda_{\text{max}}$  262 nm ( $\epsilon=10,830$ ), shoulder 280 nm

#### Method B:

1-Methyl-9-(1,3-dioctanoyloxy-2-propoxymethyl) guanine (410 mg, 0.79 mmol) and methyl iodide (227 mg, 1.6 mmol) were mixed in dry DMF (4 ml) and stirred in a pressure vessel at  $70^\circ$  for 6 hours. The reaction mixture was evaporated to dryness and the oil so formed was dissolved in  $\text{CHCl}_3$  and ethyl ether was added by diffusion. Slightly colored product (430 mg, 82% yield) was obtained which was recrystallized to give material identical to that prepared by Method A.

#### EXAMPLE 14

7-Methyl-9-(1,3-dihydroxy-2-propoxymethyl)guanine cyclic monophosphate

See C. B. Reese and J. E. Sulston, Biochem. Biophys Acta 149, 293 (1967) who use a similar method for methylation of guanine-containing dinucleotides.

9-(1,3-Dihydroxy-2-propoxymethyl)guanine cyclic monophosphate, sodium salt (0.45 mmol) is dissolved in  $\text{H}_2\text{O}$  (75 ml) and to the stirred solution is added dimethyl sulfate (2.0 g). The pH is maintained at 5.5 by the dropwise addition of 0.5 M aqueous KOH. After 2 hours, an additional 2.0 g of

dimethyl sulfate is added and after a further 6 hours of reaction the solution is extracted with Et<sub>2</sub>O (2 x 100 ml) and the aqueous phase is concentrated to small volume. This is then applied to a Dowex 1 x 2 (Cl<sup>-</sup> form) ion-exchange column, packed and developed in H<sub>2</sub>O. The product is eluted just after the solvent front and fractions containing the title compound are pooled and evaporated to dryness. This material is dissolved in a little H<sub>2</sub>O and lyophilized to give the product as a white powder.

#### EXAMPLE 15

1,7-Dimethyl-9-(1,3-dihydroxy-2-propoxymethyl)guanine cyclic monophosphate

##### Method A:

9-(1,3-Dihydroxy-2-propoxymethyl)guanine cyclic monophosphate, sodium salt is methylated in DMSO in the presence of K<sub>2</sub>CO<sub>3</sub> (3.5 molar equivalents) and methyl iodide (3.5 molar equivalents) as described in Example 7 (Method A). The crude phosphotriester product is hydrolyzed with dilute acid and the title compound is purified by passage down a Dowex 1x2 (Cl<sup>-</sup> form) ion-exchange column as described in Example 14.

##### Method B:

1-Methyl-9-(1,3-dihydroxy-2-propoxymethyl)-guanine cyclic monophosphate, sodium salt is methylated in H<sub>2</sub>O with dimethyl sulfate as described in Example 14 to give 1,7-dimethyl-9-(1,3-dihydroxy-2-propoxymethyl)guanine cyclic monophosphate.

EXAMPLE 11Oil in Water Cream Base

	(S)-1,7-dimethyl-9-(2,3-dihydroxy-1-	
5	propoxymethyl)guanine iodide	5.0 g
	Lanolin, Anhydrous	20.0 g
	Polysorbate 60	4.0 g
	Sorbitan Monopalmitate	2.0 g
	Light Liquid Paraffin	4.0 g
10	Propylene Glycol	5.0 g
	Methyl Hydroxybenzoate	0.1 g
	Purified Water	to 100.0 g

EXAMPLE 12Water Soluble Ointment Base

	(S)-1,7-dimethyl-9-(2,3-dihydroxy-1-	
	propoxymethyl)guanine iodide	0.5 g
	Glycerol	15.0 g
20	Macrogol 300	20.0 g
	Polyethylene Glycol 1500	64.5 g

EXAMPLE 13Tablet - (Total weight 359 mg)

25	(S)-1,7-dimethyl-9-(2,3-dihydroxy-1-	
	propoxymethyl)guanine iodide	100 mg
	Lactose	200 mg
	Starch	50 mg
30	Polyvinylpyrrolidone	5 mg
	Magnesium Stearate	4 mg

For each of Examples 11-13, combine the listed ingredients by standard techniques. Similarly.



prepare other compositions of the present invention  
by substituting other compounds of the invention  
(e.g. others of the preferred compounds disclosed on  
page 6) for (S)-1,7-dimethyl-9-(2,3-dihydroxy-  
5 1-propoxymethyl)guanine iodide.

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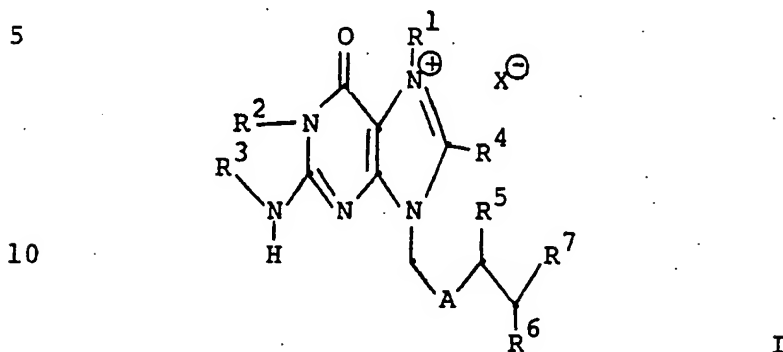
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WHAT IS CLAIMED IS:

1. A compound of the formula:



and the pharmaceutically acceptable salts thereof  
 15 wherein  $R^1$  and  $R^2$  are independently alkyl, haloalkyl, alkenyl, haloalkenyl alkynyl or haloalkynyl, each having 1 to 19 carbon atoms, or  $R^2$  is hydrogen;  $R^3$  is hydrogen, alkyl having 1 to 6 carbon atoms or hydroxyalkyl having 1 to 6 carbon  
 20 atoms;  $R^4$  is hydrogen, halogen, amino or alkyl having 1 to 4 carbon atoms;  $R^5$ ,  $R^6$  and  $R^7$  are independently selected from hydrogen, hydroxy, alkyl having 1 to 6 carbon atoms, acyloxy having 1 to 8 carbon atoms, alkoxy having 1 to 6 carbon atoms,  
 25 hydroxyalkyl having 1 to 6 carbon atoms, acyloxyalkyl having 1 to 12 carbon atoms, amino, alkylamino having 1 to 6 carbon atoms and  $-\text{PO}_3^-$  or two of  $R^5$ ,  $R^6$  and  $R^7$  taken together form a group  $-\text{OPO}_2\text{O}^-$ ,  $-\text{CH}_2\text{OPO}_2\text{O}^-$ ,  $-\text{CH}_2\text{OPO}_2\text{OPO}_2\text{O}^-$ , or  
 30  $-\text{OPO}_2\text{OPO}_2\text{O}^-$ ; A is O, S or  $\text{CH}_2$  and X is a pharmaceutically acceptable anion.

2. A compound according to Claim 1, wherein  $R^1$  is alkyl or alkenyl.

3. A compound according to Claim 1, wherein  
R<sup>1</sup> and R<sup>2</sup> are methyl, R<sup>3</sup> and R<sup>4</sup> are H, R<sup>5</sup>  
is H or hydroxymethyl, R<sup>6</sup> is H and R<sup>7</sup> is hydroxyl  
or hydroxymethyl or, alternately, R<sup>5</sup> and R<sup>7</sup> taken  
5 together are -CH<sub>2</sub>OPO<sub>2</sub>O<sup>-</sup>.

4. A compound according to Claim 1,  
wherein X is halo, alkanoate having 1 to 6 carbon  
atoms, alkylsulfonate having 1 to 6 carbon atoms,  
10 sulfate or phosphate.

5. 9-(1,3-Dihydroxy-2-propoxymethyl)-  
1,7-dimethylguaninium iodide, according to Claim 1.

15 6. 9-(1,3-Dihydroxy-2-propoxymethyl)-1,7-  
dimethylguaninium acetate, according to Claim 1.

7 9-(4-Hydroxybutyl)-1,7-dimethylguaninium  
iodide, according to Claim 1.

20

8. 9-(4-Hydroxy-3-hydroxymethylbutyl)-1,7-  
dimethylguaninium iodide, according to Claim 1.

9. 9-(2,2-dioxo-1,3,2-dioxaphosphorinan-  
25 5-yloxymethyl)-1,7-dimethylguanine, according to  
Claim 1.

10. An antiviral pharmaceutical composition  
comprising an effective amount of a compound of Claim  
30 1 and a pharmaceutically acceptable carrier.



European Patent  
Office

# EUROPEAN SEARCH REPORT

0161955  
Application number

EP 85 40 0613

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.4)
A	EP-A-0 066 208 (SYNTEX (U.S.A.) INC.) * Claims 1-7; abstract *	1,10	C 07 D 473/18 C 07 F 9/65 A 61 K 31/52 A 61 K 31/675
A	EP-A-0 074 306 (MERCK AND CO. INC.) * Claims 1-4,6,17 *	1,10	
A	EP-A-0 085 424 (SYNTEX (U.S.A.) INC.) * Claims 1-4,10,16,17; abstract *	1,10	
A	DE-A-2 539 963 (WELLCOME FOUNDATION LTD.) * Claims 1,2; examples 5,6,25; page 2, line 7 - page 3, line 2 *	1,10	
P,A	EP-A-0 130 126 (MERCK AND CO. INC.) * Claims 1,17-19 *	1,10	TECHNICAL FIELDS SEARCHED (Int. Cl.4)  C 07 D 473/00 C 07 F 9/00
A	EP-A-0 055 239 (ASTRA LÄKEMEDEL A.B.) * Claims 1,10,13,14 *	1,10	
A	DD-A- 202 717 (WELLCOME FOUNDATION LTD.) * Pages 1,2 *	1,10	
The present search report has been drawn up for all claims			
Place of search BERLIN		Date of completion of the search 28-06-1985	Examiner HASS C V F
<p><b>CATEGORY OF CITED DOCUMENTS</b></p> <p>X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document</p> <p>T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons &amp; : member of the same patent family, corresponding document</p>			



DOCUMENTS CONSIDERED TO BE RELEVANT			CLASSIFICATION OF THE APPLICATION (Int. Cl.4)
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	
A	DE-A-2 808 096 (WELLCOME FOUNDATION LTD.) * Claims 1,4; page 9 - page 10, first paragraph *	1,10	
P,A	EP-A-0 105 486 (SYNTEX (U.S.A.) INC.) * Claims 1,21 *	1,10	
			TECHNICAL FIELDS SEARCHED (Int. Cl.4)
The present search report has been drawn up for all claims			
Place of search BERLIN		Date of completion of the search 28-06-1985	Examiner HASS C V F
<b>CATEGORY OF CITED DOCUMENTS</b>			
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document			

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